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FINAL REPORT

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MOLECULAR BEAM STUDIES OF LOW ENERGY REACTIONS

PERIOD: 1 SEPTEMBER 1973-31 JANUARY 1980

ONR CONTRACT NO. NOO014-74-C-0011

PRINCIPAL INVESTIGATOR: R. H. NEYNABER

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Final Report

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Molecular Beam Studies of Low Energy Reactions
ONR Contract No. N00014-74-C-0011

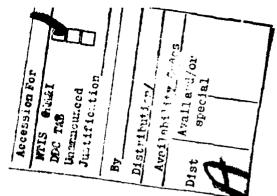
1. Contract Description

Chemi-ionization and ion-molecule reactions involving metastable and ground-state atoms are studied by both crossed and merging beams at low relative energies (i.e., 0.01 to 10 or 20 eV).

Scientific Problem

Some theories exist for chemi-ionization involving collisions of metastable and ground-state rare gases. There are very little experimental data to test these theories over a range of relative kinetic energy from 0.01 to 10 or 20 eV. We will supply such data. Theoretical work for collisions between two metastables is almost non-existent, and experimental data are scant. We will supply experimental information such as absolute and relative cross sections and branching ratios for associative to Penning ionization. This information should establish patterns to test those calculations that do exist and will stimulate further theory. Our chemiconization data also will produce some information on unknown potentials for the systems A B and C D, where A, B, C, and D are atoms and asterisks denote metastables. This information includes well depths and the dependence of the long range potential on internuclear separation.

The composition of keV neutral rare gas beams formed by charge transfer of the rare gas parent ion beam in alkalis is unknown. The beams consist of rare gas metastables (generally in two states) and ground-state atoms. The technique for generating such beams is common, and information on the composition is needed in analyzing data obtained through their use. We have developed a method for obtaining the fraction of ground-state atoms in such beams by studying appropriate ion-molecule reactions. We will apply this



method to determine unknown compositions.

No experimental information exists on low-energy resonant or near-resonant charge-transfer reactions between rare gas ions and metastables. Our experiments will supply such information. The data can be used to see if existing theories for charge transfer between ions and ground-state atoms can be extended to this case. We also will investigate energy distributions of product ions from which information on the reaction kinetics can be obtained.

3. Scientific and Technical Approach

Merging-beams techniques will be used for most of the studies. The two reactants of the process under investigation will be merged. Their velocities will be adjusted with respect to each other so that the desired relative energy in the center-of-mass system will be obtained. Product ions resulting from the reaction will be collected to give relative and absolute cross sections, and branching ratios will be obtained when appropriate.

Some crossed-beams measurements will be made of the ion-molecule reactions at relative energies above 1 eV. Again product ions will be measured to obtain cross sections.

4. Results

4.1 Chemi-ionization Studies

Our chemi-ionization studies have been largely of Penning ionization (PI) and associative ionization (AI) using merging-beams techniques. Most of the effort has been devoted to processes in which both reactants are atoms. Some of these reactions can be described as

$$R^{+} + A \rightarrow RA^{+} + e$$
 (AI) (1)

$$+ R + A^{+} + e \quad (PI) \quad , \tag{2}$$

where R^* is a metastable rare gas and A is a ground-state atom. These reactions are heteronuclear, and, hence, R is different from A. Although we have conducted some studies in which A was N or O, we will concentrate the discussion on those cases where A is a rare gas, atomic deuterium, or Na.

The remainder of the AI and PI reactions studied can be characterized as

$$R^{+} + R^{+} + RR^{+} + e$$
 (AI) (3)

$$\rightarrow R + R^{\dagger} + e \quad (PI) \tag{4}$$

$$\rightarrow R^{+} + R' + e \quad (PI) \quad , \tag{5}$$

where R^* is a metastable rare gas, and R^* can be the same as or different from R.

The R*(R'*) represents a composite of the two metastable states of the rare gas. For He the metastable states are 2^3 S and 2^1 S at energies of 19.8 and 20.6 eV, respectively. For Ne, Ar, and Kr the two metastable states are 3P_2 and 3P_0 at energies of 16.62 and 16.71 eV, 11.55 and 11.72 eV, and 9.92 and 10.56 eV, respectively.

Studies of Reactions (1)-(5) were generally made in the range $0.01 \leq w \leq 10$ eV (where W is the interaction energy, i.e., relative kinetic energy of the reactants) by measuring the product ionic currents. In addition, laboratory energy distributions of Penning ions were obtained. From these measurements the following quantities were determined: absolute and relative cross sections for AI and PI (Q_{AI} and Q_{PI}), total ionization cross sections Q_{T} ($\equiv Q_{AI} + Q_{PI}$), $r \equiv Q_{AI}/Q_{PI}$, and the branching ratio $R(\equiv Q_{AI}/Q_{T})$. In addition, information was obtained on the dynamics of the collision process and on the long-range form and the well depth ϵ^* of the potential curve $V^*(R)$ of the reactants.

An effort has been made to observe patterns in Q_{AI} , Q_{PI} , Q_{T} , and R that can be related to the attractive or repulsive nature of the reactant system. (A system is defined as attractive if $W/\epsilon^* \ll 1$, where W is in the thermal energy range, i.e., near 0.05 eV, and repulsive if $W/\epsilon^* >> 1$.) Such patterns could lead to some degree of predictability for the unmeasured Q and R of a given attractive or repulsive system.

The ideal situation would be to have a theory that could be applied to obtain dependable Q's and R's. A few theoretical models have been devised and have been applied to some systems. In their more sophisticated forms these have been difficult to use even for simple systems and, in general, require a priori knowledge of $V^*(R)$, the product potential curve $V^*(R)$, and the coupling width $\Gamma(R)$ for $V^*(R)$ and $V^*(R)$. These functions are difficult to obtain. When comparisons of Q and R can be made with experimental results over a wide energy range (e.g., $0.01 \le W \le 10$ eV), they are often disappointing. Because of these

theoretical difficulties, There is some merit in our empirical approach to estimate, at least crudely, the Q and R of a given system.

Our chemi-ionization results can best be summarized in tabular form. Shown below are Tables I and II 1 . (References in these tables are given in Ref. 1 and reproduced in this report after the presentation of the tables.) Our rare gas beams not only consisted of metastables but also ground-state species. The measured fraction of ground-states species, f_g , is given in Table I. Theoretical fractions calculated using a statistical approach are designated as f_{gt} . The ground-state species do not contribute to cheminonization.

TABLE I. Fraction f_g of ground-state species in rare-gas beams formed by charge transfer of rare-gas ions at energy E in alkali vapor at pressure p. 6

Abre gosi ^b	Alkalı	E (nev)	p ^C (millitorr)	Range of p for const fg (millitorr)	1-0	1,*	fgt f
Me	Cs	1.1	0.5	0.13-1.1	0.12	0.14	0.17
		1.6	0.6	0.5 -1.8	0.14	0 16	0.13
		4.0	0.5	0.06-2 C	0.12	C 14	0.13
Ne	Na	1.1	0 8	•••	0.05	C D6	<0.02
		1.6	0.8		0.05	C 06	<0.02
		4.0	9 0	0.2-1.5	0.05	0.06	0.03
3 _{ne}	Na	5.8	C 8	••-	0.11	0.13	0.10
Ne	No	1.39	C 48 ⁹	•••	0.45	0.49 ± 0.04	0.42
		2.75	0.65	0.04-3.7	0.53	0.57 ± 0.05	0.50
		5.5	0.62	0.23-1.5	0.53	0.57 ± 0.05	0.50
Ar	№	1.0	Ø. 15	•••	0.58	0.61 ± 0.05	0.54
		2.0	0.15		0.58	0.62 ± 0.05	0.53
		4.0	0.6	•••	0.62	0.66 ± 0.05	0.51
		5.0	0.6	0.12-1.0	0.60	0.63 ± 0.05	0.50
Kr	Cs	4.5	0.04	0.04-C 14	0.60	0.64 ± 0.07	0.60
		5.5	0.14	0.04-0.44	0.65	0.68 . 0.06	0.56

This table gives our most up-to-date values for f

Except when noted He denotes he.

Evalues are accurate to within a factor of 2. Longth of charge-transfer cell is 10 cm.

**Presumably f is constant for p less then minimum of range. **Ressumments were not made where a range is not specified except as noted by footnote g.

The $\sqrt{\gamma}+1$ for f and 1.16 for f. The γ 's are defined in Ref. 10. The f and f for we are upper bounds, see Ref. 10 for the explanation.

Time $f_{\rm gt}$ contain undesignated errors due to the use of calculated charge-transfer Qis. Pine p and E for this measurement result in multiple collisions and quenching of Relin the cell. Thus, the associated $f_{\rm ge}$ and $f_{\rm g}$ are unduly large.

TABLE II. Both for vorteus unlocular system

						61(10-16m2)*	
(Files)	faces referral	sr	c*(ev)	M(qV) or Courseignt	•	a(10 ⁻¹¹ cm ³ /sec) ^e	Nef.
₩°-Ar	Synther & Supracor	0.31-0 03		0.033	0.20-0.00	30.5 ⁺¹²⁶	B(a)
	Tang et al	0.31±0.00*	••	0.033	••	9.3 <u>e</u> 0.4 ^h	14(a),14(b
	treme et al			0.033	0.36-0 805		15
_	test et a?	••	••	435:	0.31+0.03	34.7-2.9	16
3 63)-40	Brem et al	••	••	300+	•-	10	17
	Beynater & Hagnisan	C 38-C 04		0.05	0.30±0.05	12.6-5.3	18
	Tang et al	0.36-0 007	••	0.05	<u>.</u> .	8.20-0.47	14(4),14(0
	Green et al	••	••	330s	0.32-0.905	••	15
	Bioheus	••	••	9.05	8.32 <u>+</u> 0 02	••	15
	West et al			435¢	9.30-0.04	17.6-3.5	16
p(³ p ₂ }-£r	Brow et al		••	2004		11	17
* - 3e	Noymober & Tong	0 46 <u>-</u> C 05	••	0 05	0.24+0.04	14.025.2	13
				300+		8-9.8-3.6	13
	Tong et al	0 44 <u>-</u> C 008		0.05	••	8.9+0.4 ^h	14(a),14(b)
	11langerger		••	0.05	0.20±0 07	••	\$0
	met et al	••	••	4354	0.24+0.03	34.025.2	16
r: 3 ₅₇ 2e	Bran et a'	•-	•-	200*		12	17
			••	300=		6-7 4	17
·3 _{F2} ··Re	MURSTIS et a'	••	••	3004	••	<u>4-7.5<u>-</u>1</u>	51
	Beynsper & Tons	G 34	2.1	C D÷	C.23-C 04	24 6 .5	0(b)
117 ³ 5 -#	Rorgner & Brieflaus			0.045	0.196-C-00E	33.6+€.7	22
r.2 ³ 5 -#	Fort et a'		••	£ 05	D 27+C DS	24 0-4	O(d,
112351-8	Hicker & Horper'Tr			0.05	G 16	3 5.5	B(c)
::2 3 5,-+	Hickson & Hongrer Tr	•-		C 05	0.20	34 0	B(c)
	Mates et al		2 44	••	••	••	23
	Biller & Schoefer'Tr		1.91	••	•-	••	24
	Higher et al (Thu	••	2.07	••	••		25
-80	Represent & Regnuser	0 33	0.3-C 4	0 033	£.60 <u>e</u> 0.017	\$1.2 <u>*</u> 15.5	24
· m	Reymoter et al	0. 3e	0 6	0.033	300 3±400.0	112:30	12
- 14	Corrison et a' (Tr	•-	0 56-0 6'	0 033	0 74	92	7(m)
	Brische et a'		••	0 033	0.70 <u>~</u> C 12	04 ±17	21
	Myers & Curringnur	••	••	0.033	••	101-25	26
- No	Maymager & Tang	••	0.5	6 633	40.04 <u>c</u> 0.001	197 <mark>-39</mark>	29
••	Reynober & Tanç	5 4C	0 2-0 4	0.033	0.035+C.0C*	308+6?	30

REFERENCES FOR TABLES I AND II

[From R. H. Neynaber, "Merging Beams Experiments with Excited Atoms," Electronic and Atomic Collisions, ed. by N. Oda and K. Takayanagi (North-Holland Publishing Co., Amsterdam, 1980), pp. 287-300.

- See, for example, (a) R. H. Neynaber and G. D. Magnuson, Phys. Rev. All, 865 (1975); (b) R. H. Neynaber and S. Y. Tang, J. Chem. Phys. 69, 4851 (1978); (c) A. P. Hickman and H. Morgner, J. Chem. Phys. 67, 5484 (1977); (d) J. Fort, J. J. Laucagne, A. Pesnelle, and G. Watel, Phys. Rev. A18, 2063 (1978).
- 10. R. H. Neynaber and G. D. Magnuson, J. Chem. Phys. 65, 5239 (1976) and R. H. Neynaber and S. Y. Tang, Chem. Phys. Lett. 65, 150 (1979).

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⁹A recent remodurement at 18° of G₇(and 033 of) for the his system resulted in a value educt 181 loss than she This value is larger than that in Deference 14(a) to account for a more refring value of a grow in Deference 14(a)

- 12. R. H. Neynaber, G. D. Magnuson, and S. Y. Tang, J. Chem. Phys. <u>68</u>, 5112 (1978).
- 13. R. H. Neynaber and S. Y. Tang, J. Chem. Phys. <u>70</u>, 4272 (1979).
- (a) S. Y. Tang, A. B. Marcus, and E. E. Muschlitz, Jr., J. Chem. Phys. 56, 566 (1972); (b) M. R. Woodward, R. C. Sharp, M. Seeley, and E. E. Muschlitz, Jr., J. Chem. Phys. 69, 2978 (1978).
- 15. H. L. Kramer, J. A. Herce, and E. E. Muschlitz, Jr., J. Chem. Phys. <u>56</u>, 4166 (1972).
- W. P. West, T. B. Cook, F. B. Dunning, R. D. Rundel, and R. F. Stebbings, J. Chem. Phys. <u>63</u>, 1237 (1975).
- 17. J. M. Brom, Jr., J. H. Kolts, and D. W. Setser, Chem. Phys. Lett. <u>55</u>, 44 (1978).
- 18. R. H. Neynaber and G. D. Magnuson, Phys. Rev. A14, 961 (1976).
- 19. A. Niehaus, Ber. Bunsenges. Phys. Chem. <u>77</u>, 632 (1973).
- 20. E. Illenberger, Diplom-Arbeit, Universitat Freiburg, 1971.
- D. L. Huestis, R. M. Hill, H. H. Nakano, and D. C. Lorents, J. Chem. Phys. 69, 5133 (1978).
- 22. H. Morgner and A. Niehaus, J. Phys. B12, 1805 (1979).
- 23. H. Hotop, E. Illenberger, H. Morgner, and A. Niehaus, Chem. Phys. Lett. 10, 493 (1971).
- 24. W. H. Miller and H. F. Schaefer, J. Chem. Phys. 53, 1421 (1970).
- 25. A. P. Hickman, A. D. Isaacson, and W. H. Miller, J. Chem. Phys. <u>66</u>, 1483 (1977).
- 26. R. H. Neymaber and G. D. Magnuson, J. Chem. Phys. <u>67</u>, 430 (1977).
- 27. P. Deloche, P. Monchicourt, M. Cheret, and F. Lambert, Phys. Rev. A<u>13</u>, 1140 (1976).
- 28. G. Myers and A. J. Cunningham, J. Chem. Phys. <u>67</u>, 247 (1977).
- 29. R. H. Neynaber and S. Y. Tang, J. Chem. Phys. <u>67</u>, 5619 (1977).
- 30. R. H. Neynaber and S. Y. Tang, J. Chem. Phys. <u>71</u>, 3608 (1979).

Some comments about the results in these tables are:

- a) The theoretical f in Table I are in quite good agreement with the measured values indicating the validity of predicting the populations of the rare-gas states by a statistical approach.
- b) Our experimental values of ε^* in Table II are for attractive systems and are in good agreement with the ε^* of others when comparisons can be made. Those systems with no values of ε^* are repulsive systems.
- c) Except for He*-D, the R (as seen in Table II) for attractive systems are quite small (generally < 10%) and, at thermal energies, considerably lower than the R for the repulsive systems.
- d) From Table II one notes that the Q_T for R^*-R^{**} systems [see Reactions (3), (4), and (5)] are considerably greater than those for R^*-A [see Reactions (1) and (2)] systems.

More details, additional comments, and explanations of some of the conclusions above have been presented. Furthermore, comparisons of relative Q_T versus W and R versus W for the repulsive and attractive systems in Table II have been given. One interesting observation from such comparisons is that the Q_T and R curves are nearly the same for the repulsive Ne^*-Ar , Ne^*-Kr , and Ne^*-Xe systems, whereas for attractive systems such similarities do not exist.

Chemi-ionization studies under the contract have also been done on a few systems not shown in Table II. These are the He * -H $_2$, He * -N, and He * -0 3 systems.

4.2 Ion-Molecule Studies

In addition to the chemi-ionization work the merging-beams apparatus has been used on this contract to study the charge transfer between $\operatorname{He}^+(1S)$ and Ne^+ in the energy range $0.1 \le W \le 500$ eV. The cross section monotonically increases with W. The threshold for the reaction is near 0.1 eV. A modified Demkov approach is used to calculate cross sections, which agree very roughly with the experimental values above $W \approx 3$ eV. At low W the agreement is poor.

A crossed-beams study was also done on reactions of C^+ and O^+ with CO_2 . The energy range of C^+ was from 1 to 500 eV.

5. Publications

Publications credited to the contract are listed below.

- a. R. H. Neynaber and G. D. Magnuson, "Associative and Rearrangement Ionization in Collisions of Metastable Helium with H₂," J. Chem. Phys. <u>61</u>, 749 (1974).
- b. R. H. Neynaber and G. D. Magnuson, "Product Energies for $N_2^+ + O_2 \rightarrow N_2 + O_2^+$," J. Chem. Phys. 61, 3490 (1974).
- c. R. H. Neynaber and G. D. Magnuson, "Chemi-ionization in Collisions of Metastable Neon with Argon," Phys. Rev. All, 865 (1975).
- d. R. H. Neynaber and G. D. Magnuson, "Associative Ionization in Collisions Between Two Excited Reactants," Phys. Rev. A12, 891 (1975).
- e. R. H. Neynaber and G. D. Magnuson, "Associative Ionization in Collisions Between Metastable Helium and Atomic Nitrogen and Oxygen," J. Chem. Phys. 64, 2840 (1976).
- f. J. A. Rutherford and D. A. Vroom, "A Study of the Reactions of C⁺ and O⁺ with Carbon Dioxide," J. Chem. Phys. 64, 3057 (1976).
- g. R. H. Neynaber and G. D. Magnuson, "Penning and Associative Ionization in the Metastable Neon-Krypton System," Phys. Rev. A14, 961 (1976).
- h. R. H. Neynaber and G. D. Magnuson, "Composition of Partially Metastable Rare Gas Beams," J. Chem. Phys. <u>65</u>, 5239 (1976).
- R. H. Neynaber and G. D. Magnuson, "Chemi-ionization in Collisions of Metastable Argon with Sodium," J. Chem. Phys. 67, 430 (1977).
- j. R. H. Neynaber and S. Y. Tang, "Chemi-ionization in Collisions of Metastable Helium with Metastable Neon," J. Chem. Phys. 67, 5619 (1977).
- k. R. H. Neynaber, G. D. Magnuson, and S. Y. Tang, "Chemi-ionization in Collisions of Metastable Helium with Metastable Helium," J. Chem. Phys. 68, 5112 (1978).
- 1. S. Y. Tang and R. H. Neynaber, "Charge Transfer Between Helium Ions and Metastable Neon," Phys. Rev. A18, 1925 (1978).
- m. R. H. Neynaber and S. Y. Tang, "Penning and Associative Ionization in the Metastable Helium-Atomic Deuterium System," J. Chem. Phys. 69, 4851 (1978).
- n. R. H. Neynaber and S. Y. Tang, "Chemi-ionization in Collisions of Metastable Neon with Xenon," J. Chem. Phys. <u>70</u>, 4272 (1979).
- o. R. H. Neynaber and S. Y. Tang, "Composition of Partially Metastable Kr Beams," Chem. Phys. Lett. <u>65</u>, 150 (1979).
- p. R. H. Neynaber and S. Y. Tang, "Chemi-ionization in Collisions of Metastable Argon with Metastable Krypton," J. Chem. Phys. <u>71</u>, 3608 (1979).

6. Extenuating Circumstances

None

- 7. There were no unspent funds remaining at the end of the contract.
- 8. No graduate students or postdoctoral personnel have been associated with the contract.
- 9. During the contract, R. H. Neynaber also received partial support from the General Physics Division of the Air Force Office of Scientific Research.

References

- 1. For more details see R. H. Neynaber, "Merging-Beams Experiments with Excited Atoms," <u>Electronic and Atomic Collisions</u>, ed. by N. Oda and K. Takayanagi (North-Holland Publishing Co., Amsterdam, 1980) pp 287-300.
- 2. R. H. Neynaber and G. D. Magnuson, J. Chem. Phys. 61, 749 (1974).
- 3. R. H. Neynaber and G. D. Magnuson, J. Chem. Phys. <u>64</u>, 2840 (1976).
- 4. S. Y. Tang and R. H. Neynaber, Phys. Rev. A18, 1925 (1978).

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